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CH?CH3

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2. Complete study guide. - You will take the entire test on Monday.

Both answers should be about 6.02×10^{23} . The quantity, 6.02×10^{23} is Avogadro's constant and we call it the "mole". Just like the quantity "12" is called a "dozen", so the quantity 6.02×10^{23} is called a "mole". By definition, a mole is the quantity of atoms necessary to equal the element's atomic mass in grams.

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moles of gas in the container.) A) B) C) $(n_2)PV+T$ D) 4. Consider your answer to question 3. Compute the numerical value of the constant by putting values into the equation indicated by your answer to question 3. This constant is given a special name, "The Ideal Gas Constant" and it has the symbol "R". $= 8.31 = R$ 5.

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