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hydrogen carbonate  
decomposes on heating  
according to  $2\text{NaHCO}_3(\text{s})$   
 $\text{Na}_2\text{CO}_3(\text{s}) + \text{H}_2\text{O}(\text{g}) +$   
 $\text{CO}_2(\text{g})$  Decomposition of  
1.275 g of a moist sample of  
 $\text{NaHCO}_3$  yields a solid  
residue of 0.528 g. The mass  
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sample was

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